

Name: \_\_\_\_\_

### Homework for Lecture 4

The diagram below shows the pipeline used for emitting the exhaust gas of an engine into air. The total length of the horizontal pipe is  $l$ , and the  $\text{CO}_2$  gas produced from the engine flows into the pipe through a "T" joint, which is located  $l/5$  from left. Suppose the concentration of  $\text{CO}_2$  remains constant as  $C_0$  at the joint "T" position. After  $\text{CO}_2$  enters the pipe, it can either go left or right by taking a random walk (diffusion). Considering the small diameter of the pipe, the diffusion of  $\text{CO}_2$  can be treated as one-dimensional. When  $\text{CO}_2$  comes out of either the left or right opening, it diffuses rapidly into air, immediately reaching equilibrium with air, where the concentration of  $\text{CO}_2$  is  $0.1 C_0$ . Therefore, the flowing of  $\text{CO}_2$  in the pipe can be considered as a steady state system.

- 1). What's the concentration distribution of  $\text{CO}_2$  inside the left and right side of pipe?
- 2). What is the ratio of total amount of  $\text{CO}_2$  in the left side to that in the right side?
- 3). What is the flowing flux of  $\text{CO}_2$  inside the left and right side of pipe?

